

Replaces DVS 2205-1, Supplement 15 (September 2005)

This supplement includes reduction coefficients in the case of a media influence.

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Footnotes for Sections 1.1 and 1.2:

- 1) I: inorganic substance
O: organic substance
M: mixture of inorganic and organic substances
- 2) SS: saturated (at 20°C), aqueous solution
TP: technically pure medium
CA: commercially available composition or as occurring in nature
- 3) Not transferable to other waste waters.
- 4) 88.25 parts of water, 10 parts of sodium perchlorate, 1 part of sodium hydroxide, 0.25 parts of aniline, 0.25 parts of monochlorobenzene and 0.25 parts of toluene diamine.
- 5) On the basis of many years of experience in practice, $A_2 = 1$ (also see the DVS 2205-1 technical code, Section 4.2).
- 6) Extrapolated values according to ISO/TC 138 SC 3 N 382.
- 7) For further information, see:
Kempe, B.: Testing methods for the determination of the behaviour of polyolefins subjected to the effect of chemicals.
Materialwissenschaft und Werkstofftechnik 15 (1984), pp. 157/72.
- 8) For further information, see:
Barth, E.: The behaviour of hard PVC subjected to the effect of chemicals.
Materialwissenschaft und Werkstofftechnik 17 (1986), pp. 74-76 and 98-108.

1 Reduction coefficients A_2 in the case of a media influence for PE, PP and PVC

1.1 Media with stress-independent A_2 values

Medium	Chemical formula	1) Concentra- tion ²⁾ %	PE-HD				PP				PVC-NI			
			20°C	40°C	60°C	80°C	20°C	40°C	60°C	80°C	95°C	20°C	40°C	60°C
Air	O ₂ N ₂	I 100	1	1	1				1	1	1	1	1	1
Alkaline solutions ⁴⁾		M 100				2								
Aluminium chloride ⁵⁾	AlCl ₃	I ≤ SS	1	1	1		1	1	1			1	1	1
Aluminium sulphate ⁵⁾	Al ₂ (SO ₄)	I ≤ SS	1	1	1		1	1	1			1	1	1
Alums (Me(I)/Me(III) sulphates) ⁵⁾		I ≤ SS	1	1	1		1	1	1			1	1	1

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DVS, Technical Committee Working Group "Joining of Plastics"

Medium	Chemical formula	1)	Concentra-tion ²⁾ %	PE-HD				PP					PVC-NI		
				20°C	40°C	60°C	80°C	20°C	40°C	60°C	80°C	95°C	20°C	40°C	60°C
Ammonia liquor ⁵⁾	NH ₄ OH	I	≤ SS	1	1	1		1	1	1			1	1	1
Ammonia, gaseous ⁵⁾	NH ₃	I	TP	1	1	1		1	1	1			1	1	1
Ammonia, liquid ⁵⁾	NH ₃	I	TP	1	1	1		1							
Ammonia solution ⁵⁾	NH ₄ OH	I	≤ SS	1	1	1		1	1	1			1	1	1
			15									1.67			
			≤ 30										1	1	1
Ammonium acetate ⁵⁾	CH ₃ COONH ₄	M	≤ SS	1	1	1		1	1	1					
Ammonium bromide ⁵⁾	NH ₄ Br	I	≤ SS	1	1	1		1	1	1			1	1	1
Ammonium carbonate ⁵⁾	(NH ₄) ₂ CO ₃	I	≤ SS	1	1	1		1	1	1			1	1	1
Ammonium chloride ⁵⁾	NH ₄ Cl	I	≤ SS	1	1	1		1	1	1			1	1	1
Ammonium fluoride ⁵⁾	NH ₄ F	I	> 10	1	1	1		1	1	1			1	1	1
Ammonium hydro-gencarbonate ⁵⁾	(NH ₄)HCO ₃	I	≤ SS	1	1	1		1	1	1			1	1	
Ammonium nitrate ⁵⁾	NH ₄ NO ₃	I	≤ SS	1	1	1		1	1	1	1	1	1		
Ammonium sulphate ⁵⁾	(NH ₄) ₂ SO ₄	I	≤ SS	1	1	1		1	1	1	1	1	1	1	1
Ammonium sulphide ⁵⁾	(NH ₄) ₂ S	I	≤ SS	1	1	1		1	1	1				1	1
Ammonium-dihydrogen-phosphate ⁵⁾	NH ₄ H ₂ PO ₄	I	≤ SS	1	1	1		1	1	1	1	1	1	1	1
Amyl alcohol ⁸⁾	C ₅ H ₁₁ -OH	O	TP												2.36
Apple juice ⁵⁾		O	CA	1	1	1		1	1	1	1	1	1	1	1
Arkopal N 100 ⁸⁾		O	2												1.28
			5												2.17
Barium carbonate ⁵⁾	BaCO ₃	I	≤ SS	1	1	1		1	1	1			1	1	1
Barium chloride ⁵⁾	BaCl ₂	I	≤ SS	1	1	1		1	1	1			1	1	1
Barium hydroxide ⁵⁾	Ba(OH) ₂	I	≤ SS	1	1	1		1	1	1	1	1	1	1	1
Barium nitrate ⁵⁾	Ba(NO ₃) ₂	I	≤ SS	1	1	1		1	1	1			1	1	1
Barium salts ⁵⁾		I/M	≤ SS	1	1	1		1	1	1	1	1	1	1	1
Barium sulphate ⁵⁾	BaSO ₄	I	≤ SS	1	1	1		1	1	1			1	1	1
Barium sulphide ⁵⁾	BaS	I	≤ SS	1	1	1		1	1	1			1	1	1
Battery acid (see sulphuric acid)	H ₂ SO ₄	I	≤ 51		1			1	1	1			1	1	1
Beer ⁵⁾		O	CA	1	1	1		1	1	1	1	1	1	1	1
Benzene sulphonic acid ⁵⁾		O	40										1	1	1
Benzyl alcohol ⁸⁾	C ₆ H ₅ CH ₂ OH	O	TP												4.54
Boric acid ⁵⁾		I											1	1	1
Butanol – butyl alcohol ⁸⁾	C ₄ H ₉ OH	O	TP												2.5
Buttermilk ⁵⁾		O	CA	1	1	1		1	1	1	1	1	1	1	1
Cadmium chloride ⁵⁾	CdCl ₂	I	≤ SS	1	1	1		1	1	1			1	1	1

Ansicht des Regels